



Room temperature multiferroism in CaTcO_3 by interface engineering



Hongwei Wang^{a,b}, Lixin He^{a,*}, Xifan Wu^{b,*}

^a Key Laboratory of Quantum Information, University of Science and Technology of China, Hefei, Anhui 230026, People's Republic of China

^b Department of Physics, Temple University, Philadelphia, PA 19122, USA

ARTICLE INFO

Article history:

Received 1 July 2014

Received in revised form 20 August 2014

Accepted 13 September 2014

Available online 4 October 2014

Keywords:

Multiferroics

Superlattice

Room temperature

Interface engineering

ABSTRACT

We carry out systematic studies of the structural instabilities of the new perovskite family ATcO_3 ($A = \text{Ca}, \text{Sr}, \text{Ba}$) by first-principles calculations. The orthorhombic ground state of CaTcO_3 and SrTcO_3 are found to be closely related to structural distortions due to the zone boundary phonon instabilities. We also identify a weak ferroelectric instability in CaTcO_3 , which is however suppressed at the ground state due to the large antiferrodistortive mode. We show that the ferroelectricity in CaTcO_3 can be induced in the $\text{CaTcO}_3/\text{BaTcO}_3$ superlattices. Two types of interface mechanisms are involved, one is realized by the mismatch of the antipolar modes between two parent bulk materials and the other is the suppression of antiferrodistortive mode at interface. The Néel temperature of $\text{CaTcO}_3/\text{BaTcO}_3$ superlattices is found to be ~ 816 K, indicating that CaTcO_3 can be engineered into a new room temperature multiferroic material.

© 2014 Elsevier B.V. All rights reserved.

1. Introduction

ABO_3 perovskite continues to prove itself to be an important family of multifunctional materials. This is because various instabilities can be simultaneously present within its simple cubic structure at high temperature [1–8]. Many of these instabilities such as magnetic, ferroelectric, antiferrodistortive, and antiferroelectric orderings and their interactions are closely associated with different functionalities that are useful for the device applications [9]. Recently a lot of excitement has been generated, again, in ATcO_3 ($A = \text{Ca}, \text{Sr}, \text{Ba}$) as a new family of perovskite [10–12]. Similar to its isovalent neighbor of Mn ($4s^2 3d^5$), technetium has a $5s^2 4d^5$ electronic configuration with a occupied t_{2g} and empty e_g bands according to the crystal field theory. ATcO_3 has the G-type antiferromagnetic (AFM) configuration experimentally [10,12]. Strikingly, ATcO_3 is discovered to have an anomalously high magnetic ordering temperature, e.g., the Néel temperature of CaTcO_3 is approximately 800 K [10–12], which adopts Pnma space group symmetry all the way below 1273 K [10]. This is in sharp contrast to that of manganese perovskite, in which CaMnO_3 only has a Néel temperature of 123 K [13,14].

Progress has already been made in understanding the electronic structures and the magnetic exchange properties in ATcO_3 ($A = \text{Ca}, \text{Sr}, \text{Ba}$) [10–12]. However, the detailed studies of the structural instabilities and their interactions are not yet available in litera-

ture. More importantly, keeping this intriguingly high Néel temperature in mind, one might be wondering whether or not more functionalities could be explored in this new family of perovskite.

In this work, we first use the density functional theory (DFT) to perform a systematic investigation of the structural instabilities and distortions in ATcO_3 ($A = \text{Ca}, \text{Sr}, \text{Ba}$). A variety of structural instabilities has been identified. At ground state of CaTcO_3 and SrTcO_3 , the antiferrodistortive (AFD) modes associated with oxygen octahedral rotations are found to originate from the zone boundary phonon instabilities at M and R points. An antipolar structure is found in CaTcO_3 and SrTcO_3 originating from the antiferroelectric instability at X point and a trilinear coupling with M and R instabilities that lowers the free energy. An ferroelectric (FE) instability is discovered in only CaTcO_3 , which is however suppressed by the large AFD mode at the ground state. In the second part, we theoretically investigate the interface induced ferroelectric polarization in $\text{CaTcO}_3/\text{BaTcO}_3$ superlattice. By taking advantage of the intrinsic structural instabilities in CaTcO_3 , including antipolar and FE ones, we propose that the FE can be induced at the interfaces of the superlattice by two interface engineering approaches. First, the electric polarization can be induced by the mismatch of antipolar structure between BaTcO_3 and CaTcO_3 . Second, the large oxygen octahedral rotation will be suppressed at the interface which can be used to enhance the intrinsic FE mode [15–17]. This picture is confirmed by our DFT calculations and the analysis of $\text{CaTcO}_3/\text{BaTcO}_3$ superlattices. Finally, the Néel temperature of the superlattice is calculated to be ~ 816 K by a following Monte Carlo simulation. Our theoretical predictions clearly suggest that

* Corresponding authors.

E-mail addresses: helxu@ustc.edu.cn (L. He), xifanwu@temple.edu (X. Wu).

the room temperature multiferroism can be achieved in CaTcO_3 by interface engineering.

The rest of the paper is organized as follows. Section 2 introduces the technical details of the calculation. Section 3 is devoted into the results and discussions of structural instabilities of prototype cubic phase and orthorhombic ground state of bulk ATcO_3 ($A = \text{Ca, Sr, Ba}$), followed by describing the induced polarization in $\text{CaTcO}_3/\text{BaTcO}_3$ superlattices. We summarize in Section 4.

2. Computational methods

Our DFT calculations are done by using the VASP code package [18,19]. In particular, both hybrid functional within the Heyd–Scuseria–Ernzerhof (HSE) scheme [20] and the Perdew–Burke–Ernzerhof functional revised for solids (PBEsol) [21] within the generalized gradient approximation (GGA) are adopted to treat the exchange correlation for electrons. These DFT functionals are known to be very accurate in predicting the volume of solids which is particularly important because the FE instability is sensitive to the cell volume. A 500 eV planewave energy cutoff and $6 \times 6 \times 4$ k mesh points converge very well the results of bulk ATcO_3 ($A = \text{Ca, Sr, Ba}$). Ionic coordinates are considered to be fully relaxed until the Hellmann–Feynman force is less than 1 meV/Å. For the superlattices, we use a smaller mesh of $6 \times 6 \times 2$ k points samplings.

In our HSE hybrid functional calculations, a mixing parameter of $\alpha = 0.3$ is adopted. The adoption of mixing parameter in HSE functional is chosen because its closeness in both effective magnetic moment and lattice constant to the experimental values [10,12]. In the calculation based on PBEsol functional, an on-site Coulomb interaction term $U = 3.0$ eV and exchange interaction $J = 1$ eV are used for technetium. Our criterion of choosing the effective on-site Coulomb interaction $U - J$ is based on the matching of results in the lattice constant, band gap, and effective magnetic moment of Tc atoms to those calculated from the accurate HSE hybrid functional.

Our calculations on the cubic ATcO_3 ($A = \text{Ba, Sr, Ca}$) is based on the $\text{Pm}\bar{3}\text{m}$ space group symmetry in which the only degree of freedom that allowed to relax is the lattice constant a . In the cubic phase calculations, the main results are based on the GGA + U functional. In calculating the structural ground state of ATcO_3 ($A = \text{Ba, Sr, Ca}$), both HSE and GGA + U functionals are used, where the symmetry is fixed at the experimentally identified Pnma space group.

In the calculations of $\text{BaTcO}_3/\text{CaTcO}_3$ superlattices, HSE is only used for the $1\text{BaTcO}_3/1\text{CaTcO}_3$ short-period superlattice, whereas GGA + U is used in the other longer period superlattices. We carry out DFT calculations for $\text{BaTcO}_3/\text{CaTcO}_3$ superlattices assuming the coherent growth on the substrate of GdScO_3 ($a = 3.97$ Å) [22] and NdScO_3 ($a = 4.01$ Å) [23]. The selections of substrates introduce small lattice mismatches by about $\pm 1.26\%$ and $\pm 2.7\%$ in BaTcO_3 and CaTcO_3 respectively, which are feasible for the experimental growth. The space group symmetry of $\text{BaTcO}_3/\text{CaTcO}_3$ superlattices is found to be monoclinic (Pc). The ground state structure of superlattice is identified by systematically freezing all the structural instabilities found at cubic phase including octahedral rotations, antipolar, and ferroelectric modes into the supercell with high symmetry, in which $\sqrt{2}a \times \sqrt{2}a \times c$ is used. Then we perform the structural relaxation including both atomic positions and lattice constant along the superlattice growth direction. Finally, we have also carried out the phonon dispersion computation for the resulting superlattice structures and we found no leftover unstable mode.

To calculate the Néel temperature of the system, we perform Monte Carlo simulations [24] of a Heisenberg model on a $12 \times 12 \times 12$ cubic cell with periodic boundary conditions. Each cubic cell contains four spins. We perform the simulations at temperatures ranging from 300 to 1300 K. At each temperature we

carry out an initial 10,000 sweep to prepare the system, sample averages are accumulated over 750,000 sweeps.

3. Results and discussion

3.1. Prototype cubic perovskite structures

In perovskite, the structural instabilities of the prototype cubic phase very often give valuable insights in understanding its low temperature structures and help to analyze its functional properties such as ferroelectricity and multiferroicity [25,26]. To this end, we first relax the structural parameters at cubic phase for ATcO_3 ($A = \text{Ca, Sr, Ba}$). In particular, the structural relaxations are performed based on several hypothetical spin configurations, which are G, A, C type antiferromagnetic and ferromagnetic spin configurations respectively. The results are presented in Table 1.

It can be seen that for all the ATcO_3 under study, the G-type AFM configuration has the lowest total energy. This is also consistent with the G-type AFM magnetic configuration identified experimentally in CaTcO_3 and SrTcO_3 at room temperature [10,12]. Furthermore, The above results by GGA + U functional are checked by the more accurate HSE hybrid functional. As shown in Table 1, the results obtained from two functionals are quantitatively very close. In particular, both GGA + U and HSE accurately predict the lattice constant of cubic SrTcO_3 at 3.967 and 3.948 respectively. In comparison, the experimental value is 3.961 [27].

In Table 2, we present the Born effective charges (Z^*) and the dielectric constant (ϵ_∞) for cubic perovskite CaTcO_3 , SrTcO_3 , and BaTcO_3 . The effective charges on the A site and B site are isotropic; the effective charges of the oxygen atoms can be divided into two independent values, $Z_{\text{O}\parallel}^*$ and $Z_{\text{O}\perp}^*$, which represent the dynamic charge transfer by perturbing the oxygen atom along the O–Tc bond direction and perpendicular to it respectively. Similar to other ABO_3 perovskites, we observe anomalously large $Z_{\text{O}\parallel}^*$ resulting from the hybridization between oxygen 2p and Tc 4d electrons. However, the values are smaller than those in ATiO_3 and AMnO_3 with 3d electrons [28]. This is consistent with the trend

Table 1

Computed cubic ATcO_3 ($A = \text{Ca, Sr, Ba}$) structural parameters (Å) and total energy (eV) for different spin configurations with $\text{Pm}\bar{3}\text{m}$ symmetry. The total energies of G-type AFM structures are chosen to be the references.

		CaTcO_3	SrTcO_3	BaTcO_3
G-AFM	a_0 (Å)	3.904	3.967	4.036
	a_0 (Å)(HSE)	3.906	3.948	4.022
	Energy (eV)	0.000	0.000	0.000
A-AFM	a_0 (Å)	3.913	3.975	4.049
	Energy (eV)	1.065	1.038	0.998
C-AFM	a_0 (Å)	3.908	3.970	4.035
	Energy (eV)	0.449	0.448	0.443
FM	a_0 (Å)	3.917	3.976	4.048
	Energy (eV)	1.842	1.748	1.637

Table 2

Dielectric constants and Born effective charges (e) of cubic ATcO_3 ($A = \text{Ca, Sr, Ba}$) calculated by GGA + U functional. The structures are taken from results by GGA + U functional in Table 1.

	CaTcO_3	SrTcO_3	BaTcO_3
Z_A^*	2.69	2.66	2.87
Z_{Tc}^*	5.87	5.96	6.19
$Z_{\text{O}\parallel}^*$	−5.32	−5.38	−5.61
$Z_{\text{O}\perp}^*$	−1.62	−1.62	−1.72
ϵ_∞	8.64	8.60	9.14

of decreasing dynamic charges of B site atoms with 4d electrons in perovskite [26].

3.2. Structural instabilities at cubic phase

In Fig. 1, we report the phonon dispersion curves of the cubic ATcO₃ (A = Ca, Sr, Ba) along the high symmetry lines of the Brillouin zone. The lowest phonon frequencies at the high symmetry points (Γ , X, M, R) points are also listed in Table 3. In order to check the results by GGA + U, we also report the phonon frequencies by HSE hybrid functional in Table 3. It can be seen that, GGA + U does give qualitatively consistent features compared to those computed from HSE functional [29].

Compared to the structural instabilities in SrTcO₃ and CaTcO₃, all phonon modes in BaTcO₃ are stable. Obviously, in both SrTcO₃ and CaTcO₃, the largest unstable phonons are M_3^+ and R_5^- modes, which can be identified by the largest imaginary frequencies

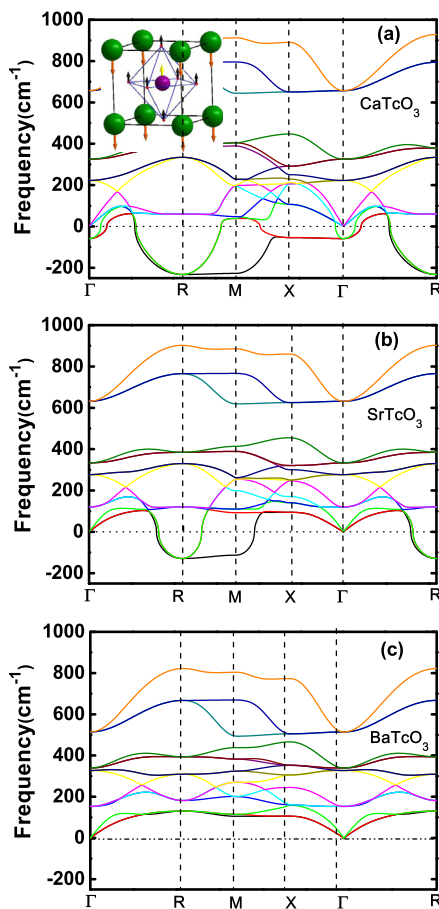


Fig. 1. Phonon dispersions of the cubic ATcO₃ (A = Ca, Sr, Ba) along the high-symmetry points, which are computed by GGA + U functional.

Table 3

Computed lowest phonon frequencies (cm⁻¹) of cubic ATcO₃ (A = Ca, Sr, Ba) by HSE hybrid functional.

Approach	Material	Γ_4^-	R_5^-	M_3^+	X_5^-
HSE	CaTcO ₃	58.28i	232.39i	232.46i	53.34i
	SrTcO ₃	118.50	127.81i	109.73i	94.45
	BaTcO ₃	151.51	130.44	106.39	103.41
GGA + U	CaTcO ₃	88.06i	221.82i	216.48i	81.39i
	SrTcO ₃	134.09	152.11i	137.43i	119.42
	BaTcO ₃	214.82	141.43	167.12	141.43

around 200i cm⁻¹. These large R and M points unstable modes suggest a strong tendency of out-of-phase and in-phase antiferrodistortive (AFD) oxygen octahedral rotation respectively.

The structural distortions due to the M_3^+ and R_5^- instabilities will be discussed in the next subsection. They give rise to the experimentally identified orthorhombic ground state of Pnma space group symmetry. Besides the AFD instabilities, interestingly, a structural instability X_5^- is also identified at Brillouin zone boundary point (X) in CaTcO₃. X_5^- mode in CaTcO₃ has a smaller imaginary frequency at 53i cm⁻¹. The X_5^- mode is an antipolar structural instability and its eigenvector can be described by the Ca and O atoms move in the opposite directions in the CaO layers. The X_5^- structural distortion is also developed at Pnma ground state in CaTcO₃ and SrTcO₃, which will also be discussed in the next subsection.

However, CaTcO₃ exhibits another weak unstable phonon mode Γ_4^- at Brillouin zone center. The Γ_4^- mode is a polar phonon mode, which may generate the electric polarization. The eigenvector of Γ_4^- phonon is depicted in the insert of Fig. 1(a), which can be described by the relative displacement of oxygen octahedral cage and the Ca atom. This particular pattern of ferroelectric phonon instability is referred as the *last* mode in literature [30].

This weak FE instability only exists in CaTcO₃ and is absent in either BaTcO₃ or SrTcO₃. It is well-known that in the conventional perovskite ATiO₃ (A = Ba, Sr, Ca), the cubic BaTiO₃ has the strongest FE instability and the largest spontaneous polarization at low temperature tetragonal and rhombohedral phases [31]. On the other hand, cubic CaTiO₃ has the weakest FE polarization and stays as a paraelectric material at ground state. However, this can be understood by Goldschmidt criterion in terms of the so-called tolerance factor $t = \frac{r_A + r_O}{\sqrt{2}(r_B + r_O)}$ [32]. For comparison, we present the tolerance factor for ABO₃ families with B = Ti, Mn, and Tc respectively in Table 4. According to the Goldsmiths criterion, when $t = 1$, the perovskite prefers the idea cubic structure; when $t > 1$ it prefers the B-site driven ferroelectricity; when $t < 1$ the perovskite prefers both the A-site driven ferroelectricity and oxygen octahedral rotation at the same time. Because the radius of Tc is larger than that of Ti, the tolerance factor in BaTcO₃ is very close to 1 favoring the idea cubic structure; on the other hand, the tolerance factor of BaTiO₃ is larger than 1 favoring the B-site driven ferroelectricity. This is consistent with the absence of any structural instability in BaTcO₃.

Similar weak FE instabilities are also found in other perovskite materials, such as CaMnO₃ [33] and CaTiO₃ [34]. However, the electric polarization is not able to develop in these materials because they are suppressed by the large oxygen octahedral rotation AFD modes, which is in strong competition with FE modes. In Section 3.4, we will discuss how to induce the FE by interface engineering method in superlattices.

3.3. Pnma ground state

In experiments, the only available low symmetry structures found in CaTcO₃ and SrTcO₃ are the orthorhombic structures with Pnma space group symmetry [10,12]. The structural distortions from cubic to orthorhombic phase can be viewed as the combined condensations of the R_5^- and M_3^+ zone boundary unstable phonon modes. As a result, in the Pnma structure, the oxygen octahedral exhibits out-of-phase rotation around [110] axis and in-phase oxygen rotation around [001] axis. In Glazer notation [35], they are referred to as $a^-a^-c^0$ and $a^0a^0c^+$ type oxygen rotations respectively. Following the convention in literature [36], we refer the oxygen octahedral rotation around [001] as *rotation* and refer oxygen octahedral rotation around [001] axis as *tilting* in the following paragraphs.

Table 4

Computed tolerance factors (t) of typical perovskite-type oxides ABO_3 (with $A = \text{Ba, Sr, Ca}$ and $B = \text{Ti, Mn, Tc}$).

t	CaTcO ₃ 0.914	SrTcO ₃ 0.947	BaTcO ₃ 1.004
t	CaTiO ₃ 0.966	SrTiO ₃ 1.002	BaTiO ₃ 1.062
t	CaMnO ₃ 0.959	SrMnO ₃ 0.994	BaMnO ₃ 1.054

In Table 5, we present the relaxed orthorhombic structure of ATcO₃ ($A = \text{Ba, Sr, Ca}$) with Pnma space group symmetry, in which the available experimental structures are also presented for comparison. It can be seen that the HSE hybrid functional accurately reproduces the experimental structures. On the other hand, the PBEsol functional slightly overestimates the lattice constants and volumes, which are typical for the GGA approximation of exchange correlation functionals.

It is worth to note that with the increasing radius of A site atoms, the rotational and tilting angles are decreasing from very large angle in CaTcO₃ to almost zero in BaTcO₃ as shown in Table 5. At the same time, the crystal structures also change from being strongly orthorhombic in CaTcO₃ to being cubic symmetry in BaTcO₃. Indeed, the above trend is consistent with the Goldschmidt criterion in terms of the increasing structural factors as we have discussed in the previous subsection.

Except for the zone boundary R_5^- and M_3^+ structural modes that generate the Pnma ground state structure, it is also interesting to note that the other less unstable structural instability, X_5^- , is also condensed in orthorhombic CaTcO₃ and SrTcO₃ generating the antipolar structure. This induced antipolar structure can be understood by the trilinear coupling term $\sim \lambda Q_{R_5^-} Q_{M_3^+} Q_{X_5^-}$ in the free energy phenomenologically [37,38]. As shown in Fig. 2(a), when only X_5^- structural distortion is frozen into the CaTcO₃, the total energy starts with a shallow double-well potential. This is consistent with the weak X_5^- phonon instability as discussed in previous subsection. However, when both R_5^- and M_3^+ structural distortions are present, the resulting energy is much lower. The resulting single-well potential profile indicates that the antipolar structure is induced improperly by the trilinear coupling in the free energy term, as shown in Fig. 2(d).

Unfortunately, the FE structural instability at cubic phase although has similar imaginary frequency to X_5^- , is not condensed at Pnma ground state. This is due to the strong competition between FE and oxygen octahedral rotation [15,39]. In the end, we have recalculated the phonon dispersion for all the ATcO₃ materials in the orthorhombic phase and no unstable mode is found any more indicating that the we have found the ground state structures theoretically.

3.4. Induced ferroelectricity by interface design in proposed BaTcO₃/CaTcO₃ superlattice

In this subsection, we focus on the discussion of the induced FE in CaTcO₃ by possible material engineering methods. In particular,

Table 5

Theoretically computed structural parameters, magnetic exchange angles ($^\circ$) [11], oxygen octahedral rotation angles ϕ_r ($^\circ$), and tilting angles ϕ_t ($^\circ$) of ground state ATcO₃ ($A = \text{Ca, Sr, Ba}$) with Pnma symmetry. The available experimental values [11] are also shown in parenthesis.

		a(Å)	b(Å)	c(Å)	V(Å ³)	Tc – \widehat{O}_1 – Tc	Tc – \widehat{O}_2 – Tc	ϕ_r	ϕ_t
Pnma (HSE)	CaTcO ₃	5.53 (5.53)	7.71 (7.70)	5.39 (5.39)	230.05 (229.15)	150.90 (150.43)	150.91 (151.53)	10.04	14.55
	SrTcO ₃	5.56 (5.54)	7.86 (7.85)	5.61 (5.58)	244.92 (242.74)	163.39 (161.57)	167.76 (166.96)	0.60	8.32
	BaTcO ₃	5.69	8.05	5.67	260.20	179.96	179.90	0.049	0.019
Pnma (GGA + U)	CaTcO ₃	5.57 (5.53)	7.73 (7.70)	5.41 (5.39)	232.08 (229.15)	148.41 (150.43)	148.91 (151.53)	10.63	15.80
	SrTcO ₃	5.59 (5.54)	7.90 (7.85)	5.60 (5.58)	247.51 (242.74)	161.13 (161.57)	161.24 (166.96)	6.18	9.13
	BaTcO ₃	5.71	8.08	5.71	262.87	179.94	179.88	0.058	0.032

we are going to take advantage of two intrinsic structural instabilities in cubic CaTcO₃, i.e. X_5^- and Γ_4^- , which are antipolar and polar modes respectively. The antipolar structural distortion is present at the orthorhombic ground state, while the polar one is not due to the large AFD octahedral rotations. With the intriguingly high Néel temperature found in CaTcO₃, it is particularly desirable that the FE polarization could be recovered and turned into a new room temperature multiferroic material.

The above can be achieved in the artificial materials based on modern thin-film technology. In general, there are two approaches, which are the *epitaxial strain* and *interface engineering* methods respectively. We first try the epitaxial strain method. By using the coupling between polar instability and epitaxial strain applied by the substrate, the epitaxial strain has been very often used to induce the FE in other perovskite materials [40]. To do so, we apply both tensile and compressive strain on the cubic CaTcO₃ systematically in the range of $\pm \sim 5\%$. At the same time, we lower the space group symmetry in the calculations to allow the development of polarization along both c axis and in the ab plane. Unfortunately, our calculations show that CaTcO₃ persists to be paraelectric even as much as $\pm 5\%$ has been applied.

We then focus on the possibility of interface engineering. Compared with epitaxial strain, more dramatic changes in the strengths of instabilities can be achieved at the interfaces resulting in an overall change of functionalities [15–17,41]. We propose that the FE can be induced by the interfaces of the CaTcO₃/BaTcO₃ superlattices. In particular, two types of interface engineering mechanisms are expected to take place that have been demonstrated recently in the literature [15,42]. First, FE can be induced by the mismatch of antipolar instabilities of two parent Pnma perovskites. This mismatch brings out a finite polarization at the interface due to the non-terminating cancellation of antipolar moment. In literature, this is the so-called *improper* induced FE, because the antipolar in bulk is induced by the other two structural distortions R_5^- and M_3^+ according to the trilinear coupling term $\sim \lambda Q_{R_5^-} Q_{M_3^+} Q_{X_5^-}$ in free energy as discussed in the previous subsection [37,38]. With a large antipolar structural distortion in CaTcO₃ and an almost cubic BaTcO₃, it can be expected that the electric dipole moment can be maximized at the interface of CaTcO₃/BaTcO₃ superlattices. Secondly, it has been recently shown that a substantial reduction of AFD associated with octahedral rotations at the interface layer of CaTiO₃/BaTiO₃ can be used to enhance the FE [15]. This is the so-called induced *proper* FE since it originates from the softening of intrinsic polar instability, e.g. Γ_4^- in cubic perovskite. Based on the same reasoning, we expect that the similar reduction of AFD mode will occur at the interface of CaTcO₃/BaTcO₃, which will further enhance the polarization. As a result, it will be very interesting to check if FE can be induced in the CaTcO₃/BaTcO₃ short-period superlattices by the above comprehensive interface engineering mechanisms.

We assume the coherent growth of BaTcO₃/CaTcO₃ superlattices on the substrate of GdScO₃ as schematically shown in Fig. 3. The first principles' results are summarized in Table 6. As one of the key results, the 1BaTcO₃/1CaTcO₃ superlattice does exhibit a FE ordering with a spontaneous polarization $P_s = 5.92 \mu\text{C}/\text{cm}^2$ by HSE func-

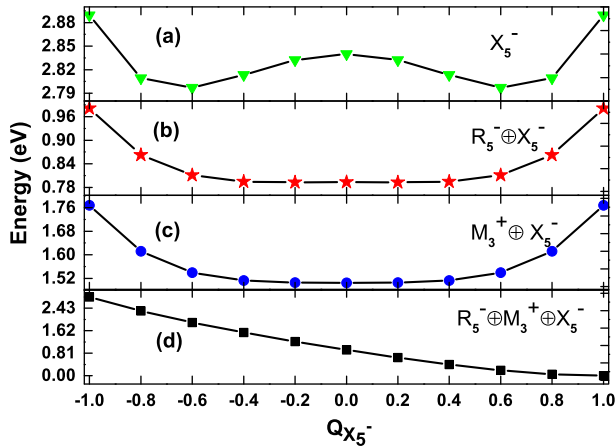


Fig. 2. Energies profiles in CaTcO₃ as functions of mode amplitudes with frozen (a) $Q_{X_5^-}$ only, (b) both $Q_{R_5^-}$ and ground state mode amplitude of $Q_{X_5^-}$, (c) both $Q_{X_5^-}$ and ground state mode amplitude of $Q_{M_3^+}$, (d) $Q_{X_5^-}$, and ground state mode amplitudes of $Q_{R_5^-}$, $Q_{M_3^+}$. The normalization factor 1 is chosen by the mode amplitude (Q) at relaxed Pnma ground state in Table 5.

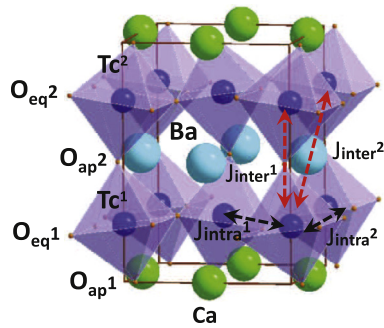


Fig. 3. Schematic atomic structure of relaxed 1CaTcO₃/1BaTcO₃ superlattice at theoretical DFT ground state with monoclinic space group symmetry Pc.

tional (or similarly $6.34 \mu\text{C}/\text{cm}^2$ by PBEsol functional) along the [110] in-plane axis. On the other hand, the pure CaTcO₃ grown on the same substrate does not exhibit any FE structural instability.

The emergence of FE ordering only in superlattice suggests that it is an interface effect. In order to further elucidate its interfacial origin, we employ the layer polarization decomposition to analyze the local polarization profile based on linear approximation [28] as shown in Table 6. In strained bulk CaTcO₃, the local polarization profile is dominated by the clear antipolar ordering [42]. The relative displacement of Ca and O atom in the same plane generates a large dipole along [110] and $[\bar{1}\bar{1}0]$ directions in alternate layers resulting a zero total polarization [42]. In superlattices, this large antipolar type local polarization $\sim 27.6 \mu\text{C}/\text{cm}^2$ is preserved in the chemical environment, in which CaO is sandwiched by two TcO₆ octahedra. However, its magnitude is also perturbed at the interfaces as shown in Table 6. The local polarizations of CaO layer in 3:3 superlattice can be divided into two categories, bulk p_{CTO} and interfaces of type I (p_{in}^{I}), which is defined by the local polarization of a CaO layer with one BaO layer as neighbor. One can see that the p_{CTO} in 3:3 superlattice is almost the same value as that in strained bulk CaTcO₃. The p_{in}^{I} is smaller than p_b , which is due to the suppressed oxygen rotations and tilting at the interface. In 2:2 superlattice, there is only one type of local polarization profile of CaO layer which belongs to interface type-I and p_{in}^{I} has a similar magnitude as that of 3:3 superlattice indicating the same chemical environment. However, in 1:1 superlattice, CaO layer is dominated by the type-II interfaces, in which the CaO layer is sandwiched by

two BaO layers. This is in qualitatively difference with interface of longer period superlattices resulting in that $p_{\text{in}}^{\text{II}} < p_{\text{in}}^{\text{I}}$. A question arises: how we could understand the two different local polarization of interfacial CaO layers in one unified picture? We describe the interface effect on the local polarization by perturbation theory. It can be further assumed that the magnitude of perturbation is within the linear regime, in which $p_{\text{in}}^{\text{II}}$ (p_{in}^{I}) are only different from the bulk value by the perturbations introduced by the suppressed AFE ordering from one (two) neighboring BaO layers. Thus, one can arrive at $p_{\text{in}}^{\text{II}} = p_{\text{CTO}} - 2 \times (p_{\text{CTO}} - p_{\text{in}}^{\text{I}}) \sim 19 \mu\text{C}/\text{cm}^2$. This is very close $19.38 \mu\text{C}/\text{cm}^2$ obtained from the actual DFT calculation.

The above local profile analysis clearly indicates that a large part of the interface induced FE is from the mismatch of antipolar distortion (X_5^- mode). The X_5^- phonon mode can be described as the atomic displacements along [110] direction including only the A site atoms and the apical oxygen atoms O_{ap} in the same layer. However, the local analysis also shows that the AFD is largely suppressed at the interfaces. Therefore, a proper FE contribution can be also expected from the induced polarization. The proper FE originates from the zone center Γ_4^- in cubic perovskite and transforms into Γ_5^- with lower space group symmetry Pc in the superlattices. The displacement of proper FE can be computed by the first-principles calculations in 1:1 superlattice with space group symmetry P4mm, in which the octahedral rotation and tilting are completely eliminated. As shown in Table 7, the proper FE mode displacement includes all the O_{ap} , equatorial oxygen O_{eq} , and A site atoms moving along [110] directions. Based on the modes's difference between improper and proper, we then performed the polarization mode decompositions into proper and improper contributions as shown in Table 7. Indeed, it can be seen that the improper polarization comprises the major part (78%) of the induced polarization, while the proper one contributes to the 22% rest of induced polarization. It can also be seen that the M_5^- and the M_3^+ representing the $a^0a^0c^-$ and $a^-a^-c^0$ type oxygen octahedral rotations do not have any contribution to induced polarization. This is consistent with the AFD nature of these oxygen rotations. The induced proper ferroelectricity is of particular interest. This is because it is associated with Tc ions, which are also responsible for the magnetism in these materials. Therefore, it is possible to achieve strong magnetoelectric coupling in the CaTcO₃/BaTcO₃ short-period superlattices [16,17].

Interestingly, with both the proper and improper polarization at the interface, it is possible that the proper polarization can be further engineered by the epitaxial strain. In perovskite materials, such as SrTiO₃ and CaTiO₃, the [110] in-plane polarization can be increased by applying tensile epitaxial strains. As shown in Table 7, we report the same mode decomposition of induced polarization for a larger tensile strain on NdScO₃ substrate. It can be seen that under this additional tensile strain, the displacement of equatorial oxygen O_{eq} is greatly increased. As a result, the induced proper polarization is also magnified by a factor of two. At the same time, we see a reduction of the improper contribution to the induced polarization. This is due to the smaller in oxygen octahedral rotation and tilting angles. It should be reminded that the effect of strain engineering is absent in bulk CaTcO₃. This additional strain tunability of polarization further confirm that the interface induced polarization is of both proper and improper properties. Since the polarization of $n\text{BaTcO}_3/n\text{CaTcO}_3$ is induced by interface, the maximum polarization is achieved in the shortest period superlattice, i.e. 1BaTcO₃/1CaTcO₃, in which the interfaces are the densest.

Both BaTcO₃ and CaTcO₃ have been found to have high Néel temperature well above room temperature. It has been argued that the more delocalized 4d orbital enhance the covalent hybridization with neighboring oxygen atoms, which in turn increase the hopping matrix according to the Anderson–Goodenough–Kanamori rules [10,44]. Therefore, it can be expected that the large magnetic

Table 6
Layer-by-layer decompositions of polarization p ($\mu\text{C}/\text{cm}^2$), oxygen octahedral rotation angles ϕ_r ($^\circ$), and tilting angles ϕ_t ($^\circ$) in the superlattices of 1BaTcO₃/1CaTcO₃(1:1), 2BaTcO₃/2CaTcO₃(2:2), 3BaTcO₃/3CaTcO₃(3:3), and strained bulk CaTcO₃ (CT^S) and BaTcO₃ (BT^S). The calculations are based on by GGA + U functional. The layer polarization is based on the definition centered on AO plane [43].

	CaO _i ^I	TiO ₂	CaO _b	TiO ₂	CaO _i ^I	TiO ₂	BaO _i ^I	TiO ₂	BaO _b	TiO ₂	BaO _i ^I	TiO ₂
3:3												
p_{110}	-23.11	-	27.58	-	-23.27	-	2.61	-	-0.33	-	2.72	-
p_{001}	-5.85	-	-0.02	-	5.82	-	2.53	-	0.034	-	-2.48	-
ϕ_r	-	9.34	-	9.20	-	6.38	-	-0.24	-	-0.33	-	6.14
ϕ_t	-	15.83	-	15.82	-	9.30	-	2.78	-	2.77	-	9.30
$p_{110}^{\text{total}} = 2.28$					$p_{001}^{\text{total}} = 0.00$							
2:2												
p_{110}	-22.10	-	-	-	23.39	-	-2.85	-	-	-	2.91	-
p_{001}	-5.55	-	-	-	5.57	-	2.73	-	-	-	-2.65	-
ϕ_r	-	9.76	-	-	-	5.79	-	-	-	0.46	-	5.77
ϕ_t	-	14.62	-	-	-	9.51	-	-	-	4.38	-	9.50
$p_{110}^{\text{total}} = 0.068$					$p_{001}^{\text{total}} = 0.00$							
	CaO _i ^{II}	TiO ₂	BaO _i ^{II}	TiO ₂								
1:1												
p_{110}	19.38	-	-6.12	-	-	-	-	-	-	-	-	-
p_{001}	0.02	-	0.04	-	-	-	-	-	-	-	-	-
ϕ_r	-	5.82	-	5.94	-	-	-	-	-	-	-	-
ϕ_t	-	9.41	-	9.41	-	-	-	-	-	-	-	-
$p_{110}^{\text{total}} = 6.34$					$p_{001}^{\text{total}} = 0.03$							
	CaO _i ^{II}	-	-	BaO _i ^{II}	-	-	-	-	-	-	-	-
1:1 HSE												
p_{110}	15.60	-	-	-4.39	-	-	-	-	-	-	-	-
p_{001}	0.04	-	-	0.08	-	-	-	-	-	-	-	-
$p_{110}^{\text{total}} = 5.92$					$p_{001}^{\text{total}} = 0.06$							
	-	CaO _b	-	CaO _b	-	-	-	-	-	-	-	-
CT ^S												
p_{110}	-	27.57	-	-27.58	-	-	-	-	-	-	-	-
p_{001}	-	0	-	0	-	-	-	-	-	-	-	-
	-	BaO _b	-	BaO _b	-	-	-	-	-	-	-	-
BT ^S												
p_{110}	-	0	-	0	-	-	-	-	-	-	-	-
p_{001}	-	0	-	0	-	-	-	-	-	-	-	-

Table 7
The mode decomposition of the (CaTcO₃)₁(BaTcO₃)₁ superlattices of relaxed theoretical ground state with space group Pc grown on GdScO₃ and NdScO₃ substrates. The first-principles results are based on HSE hybrid functional.

	GdScO ₃ substrate (3.97 Å)					NdScO ₃ substrate (4.01 Å)				
	Γ_5^- (total)	Γ_5^- (proper)	Γ_5^- (improper)	M_5^-	M_3^+	Γ_5^- (total)	Γ_5^- (proper)	Γ_5^- (improper)	M_5^-	M_3^+
Ca	0.2466	0.0393	0.2073	-	-	0.2520	0.0725	0.1795	-	-
Ba	-0.0841	-0.0019	-0.0822	-	-	-0.0757	-0.0037	-0.0720	-	-
Tc ¹	-0.0034	-0.0034	-	-	-	-0.0035	-0.0035	-	-	-
Tc ²	-0.0041	-0.0041	-	-	-	-0.0035	-0.0035	-	-	-
O _{ap} ¹	-0.1367	-0.0177	-0.1190	-	0.5487	-0.1380	-0.0301	-0.1079	-	0.5635
O _{ap} ²	0.0018	0.0006	0.0012	-	0.2677	0.0042	0.0009	0.0033	-	0.2823
O _{eq} ¹	-0.0057	-0.0057	-	0.3486	0.4260	-0.0130	-0.0130	-	0.2772	0.4352
O _{eq} ²	-0.0072	-0.0072	-	0.3361	0.4270	-0.0123	-0.0123	-	0.2769	0.4357
Q _{total} (Å)	0.2944	0.0444	0.2527	0.4842	0.8582	0.2977	0.0807	0.2214	0.3918	0.8811
P ($\mu\text{C}/\text{cm}^2$)	5.92	1.31	4.61	-	-	6.70	2.79	3.91	-	-

antiferromagnetic coupling will be kept in the BaTcO₃/CaTcO₃ superlattices. To further confirm this, we map the total energies in different magnetic configurations onto the Heisenberg Hamiltonian [4]

$$H = - \sum_{ij} J_{ij} S_i \cdot S_j. \quad (1)$$

The exchange coupling constants include the nearest-neighbor (J^1) and next -nearest-neighbor (J^2), beyond which the coupling constant is found to be very small and could be neglected. This coupling constants are further decomposed into the contributions

from intralayer (J_{intra}) and interlayers (J_{inter}). The results are listed in Table 8. One can see that the exchange coupling constants are approximately the averages of those of the bulk BaTcO₃ and bulk CaTcO₃ in both intra- and inter- layer contributions. We obtain the AFM phase transition at 816 K from the Monte Carlo simulation. Furthermore, as explained in previous paragraphs, the mechanism of induced ferroelectric dipole moment is due to the non-cancellation of anti-polar moments from CaTcO₃ and BaTcO₃ at the interface layer in the superlattices. The Pnma structural phase transition temperature in CaTcO₃ has been experimentally determined to be at 1273 K [10]. The structural phase transition in BaT-

Table 8

Effective magnetic moments $m(\mu_B)$, exchange coupling constants $J(\text{meV})$, and Néel temperature $T_N(\text{K})$ of bulk CaTcO_3 , bulk BaTcO_3 , and $1\text{CaTcO}_3/1\text{BaTcO}_3$ superlattice. GGA + U functional is used in the DFT calculations.

	CaTcO_3	BaTcO_3	$1\text{CaTcO}_3/1\text{BaTcO}_3$
m	2.14	2.09	2.05
J_{inter}^1	−38.10	−69.02	−52.60
J_{intra}^1	−38.10	−69.02	−60.29
J_{inter}^2	−0.78	−1.09	−2.89
J_{intra}^2	−0.78	−1.09	−3.17
T_N	602	1089	816

$c\text{O}_3$ has not been reported yet. This is consistent with our DFT results which show no unstable phonon mode, indicating BaTcO_3 stays cubic structure at room temperature. Therefore, the above experimental and theoretical facts suggest that $\text{CaTcO}_3/\text{BaTcO}_3$ should be multiferroic at room temperature.

4. Conclusion

Based on accurate first-principles calculations, we have performed careful studies of structural instabilities for the recently discovered perovskite ATcO_3 ($A = \text{Ba}, \text{Sr}, \text{Ca}$). The experimental structures in CaTcO_3 and SrTcO_3 are elucidated by the distortions originating from the structural instabilities and their couplings at prototype cubic phases. Based on the distinct structural instabilities in CaTcO_3 and BaTcO_3 , we propose and show that the electric polarization can be induced by the interfaces in $\text{BaTcO}_3/\text{CaTcO}_3$ superlattices, in which comprehensive interface engineering mechanisms are taken into account. The theoretically proposed superlattice is found to be a new room-temperature multiferroic and await for future experimental confirmation.

Acknowledgments

This work is supported in part by Air Force Office of Scientific Research under award numbers FA9550–13–1–0124 (X. Wu). L. He acknowledge the support from the Chinese National Fundamental Research Program 2011CB921200, National Natural Science Funds for Distinguished Young Scholars and CNSF Grant No. 11374275. X. Wu acknowledges the computational support by the National Science Foundation through TeraGrid resources provided by NICS under Grant No. [TG-DMR120045] and the National Energy Research Scientific Computing Center (NERSC).

References

- [1] H.Y. Hwang, Y. Iwasa, M. Kawasaki, B. Keimer, N. Nagaosa, Y. Tokura, *Nat. Mater.* 11 (2012) 103.

- [2] J. Hong, D. Vanderbilt, *Phys. Rev. B* 87 (2013) 064104.
 [3] J.W. Bennett, K.F. Garrity, K.M. Rabe, D. Vanderbilt, *Phys. Rev. Lett.* 110 (2013) 017603.
 [4] J.H. Lee, K.M. Rabe, *Phys. Rev. B* 84 (2011) 104440.
 [5] J.H. Lee, K.M. Rabe, *Phys. Rev. Lett.* 107 (2011) 067601.
 [6] A. Malashevich, D. Vanderbilt, *Phys. Rev. Lett.* 101 (2008) 037210.
 [7] A. Roy, M. Stengel, D. Vanderbilt, *Phys. Rev. B* 81 (2010) 014102.
 [8] J.Y. Jo, R.J. Sichel, H.N. Lee, S.M. Nakhmanson, E.M. Dufresne, P.G. Evans, *Phys. Rev. Lett.* 104 (2010) 207601.
 [9] C.H. Ahn, K.M. Rabe, J.M. Triscone, *Science* 303 (2004) 5657.
 [10] M. Avdeev, G.J. Thorogood, M.L. Carter, B.J. Kennedy, J. Ting, D.J. Singh, K.S. Wallwork, *J. Am. Chem. Soc.* 133 (2011) 1654.
 [11] C. Franchini, T. Archer, J. He, X. Chen, A. Filippetti, S. Sanvito, *Phys. Rev. B* 83 (2011) 220402.
 [12] E.E. Rodriguez, F. Poineau, A. Llobet, B.J. Kennedy, M. Avdeev, G.J. Thorogood, M.L. Carter, R. Seshadri, D.J. Singh, Anthony K. Cheetham, *Phys. Rev. Lett.* 106 (2011) 067201.
 [13] H. Taguchi, *Phys. Stat. Sol.* 88 (1985) K79.
 [14] T. Günter, E. Bousquet, A. David, Ph. Boullay, Ph. Ghosez, W. Prellier, M. Fiebig, *Phys. Rev. B* 85 (2012) 214120.
 [15] X. Wu, K.M. Rabe, D. Vanderbilt, *Phys. Rev. B* 83 (2011) 020104.
 [16] H. Wang, L. He, X. Wu, *Europhys. Lett.* 100 (2012) 17005.
 [17] C. Swartz, X. Wu, *Phys. Rev. B* 85 (2012) 054102.
 [18] G. Kresse, J. Hafner, *Phys. Rev. B* 47 (1993) R558.
 [19] G. Kresse, J. Furthmüller, *Phys. Rev. B* 54 (1996) 11169.
 [20] J. Heyd, G.E. Scuseria, M. Ernzerhof, *J. Chem. Phys.* 118 (2003) 8207.
 [21] J.P. Perdew, A. Ruzsinszky, G.I. Csonka, O.A. Vydrov, G.E. Scuseria, L.A. Constantin, X.L. Zhou, A. Burke, *Phys. Rev. Lett.* 100 (2008) 136406.
 [22] K.M. Rabe, C.H. Ahn, J.-M. Triscone (Eds.), *Physics of Ferroelectrics: A Modern Perspective*, Springer, Berlin, 2007.
 [23] R. Uecker, B. Velickov, D. Klimm, R. Bertram, M. Bernhagen, M. Rabe, M. Albrecht, R. Fornari, D.G. Schlom, *J. Crystal Growth* 310 (2008) 2649–2658.
 [24] Kun Cao, Guang-Can Guo, David Vanderbilt, Lixin He, *Phys. Rev. Lett.* 103 (2009) 257201.
 [25] J.B. Neaton, C. Ederer, U.V. Waghmare, N.A. Spaldin, K.M. Rabe, *Phys. Rev. B* 71 (2005) 014113.
 [26] S. Amisi, E. Bousquet, K. Katcho, P. Ghosez, *Phys. Rev. B* 85 (2012) 064112.
 [27] G.J. Thorogood, M. Avdeev, M.L. Carter, B.J. Kennedy, J. Ting, K.S. Wallwork, *Dalton Trans.* 40 (2011) 7228–7233.
 [28] X. Wu, O. Diéguez, K.M. Rabe, D. Vanderbilt, *Phys. Rev. Lett.* 97 (2006) 107602.
 [29] J. Hong, A. Stroppa, J. Íñiguez, S. Picozzi, D. Vanderbilt, *Phys. Rev. B* 85 (2012) 054417.
 [30] J.T. LAST, *Phys. Rev.* 105 (1957) 6.
 [31] X. Wu, D. Vanderbilt, D.R. Hamann, *Phys. Rev. B* 72 (2005) 035105.
 [32] V.M. Goldschmidt, *Die Gesetze der Kristallochemie, Die Naturwissenschaften*, 21 (1926) 477.
 [33] S. Bhattacharjee, E. Bousquet, P. Ghosez, *Phys. Rev. Lett.* 102 (2009) 117602.
 [34] C.J. Eklund, C.J. Fennie, K.M. Rabe, *Phys. Rev. B* 79 (2009) 220101. R.
 [35] A.M. Glazer, *Acta Cryst.* B28 (1972) 3384.
 [36] A.M. Glazer, *Acta Cryst.* A31 (1975) 756.
 [37] J.M. Rondinelli, C.J. Fennie, *Adv. Mater.* 24 (2012) 1961–1968.
 [38] N.A. Benedek, A.T. Mulder, C.J. Fennie, *J. Solid State Chem.* 195 (2012) 11–20.
 [39] W. Zhong, D. Vanderbilt, *Phys. Rev. Lett.* 74 (1995) 2587.
 [40] O. Diéguez, K.M. Rabe, D. Vanderbilt, *Phys. Rev. B* 72 (2005) 144101.
 [41] E. Bousquet et al., *Nature(London)* 452 (2008) 732.
 [42] A.T. Mulder, N.A. Benedek, J.M. Rondinelli, C.J. Fennie, *Adv. Funct. Mater.* 10 (2013) 4810.
 [43] B. Meyer, D. Vanderbilt, *Phys. Rev. B* 65 (2002) 104111.
 [44] J.B. Goodenough, *Magnetism and the Chemical Bond*, John Wiley and Sons, New York-London, 1993.